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## Monocillinols A and B, novel fungal metabolites from a *Monocillium* sp.

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## Abstract

Monocillinols A (1) and B (2), two novel isomeric fungal metabolites, have been isolated from the chloroform extract of the culture filtrate of a *Monocillium* sp. The structures of 1 and 2 were elucidated by interpretation of their spectral data. Published by Elsevier Science Ltd.

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Microorganisms have historically provided a rich source of structurally diverse, biologically active secondary metabolites. Previous investigations of fungi in the genus *Monocillium* have led to the isolation of a series of aromatic polyketide derivatives, several of which exhibit antifungal activity.<sup>1</sup> As part of our continuing studies of metabolites produced by microorganisms obtained from soil samples collected throughout Bangladesh,<sup>2–4</sup> we isolated the fungus *Monocillium* sp.<sup>5</sup> from soil collected in the region of Rajshahi. Purification by  $C_{18}$  HPLC of the CHCl<sub>3</sub> extract

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(80 mg) of the culture filtrate when this organism was grown in a Czapek Dox broth,<sup>6</sup> afforded two novel compounds, monocillinols A (1) (3.0 mg)<sup>‡</sup> and B (2) (2.6 mg).<sup>§</sup>

The molecular formula of monocillinol A (1), established as C<sub>11</sub>H<sub>13</sub>NO<sub>5</sub> by HR-FABMS measurements, required six double bond equivalents in the molecule. IR absorption bands at v1724 and 1667 cm<sup>-1</sup> suggested the presence of  $\alpha,\beta$ -unsaturated ester and amide carbonyl groups, respectively. The <sup>13</sup>C NMR spectrum of 1 displayed 11 carbon resonances, while DEPT and HSOC experiments revealed that seven out of the 11 carbons had attached protons. <sup>13</sup>C resonances for two carbonyls ( $\delta$  166.3 and 160.4) and four olefin carbons ( $\delta$  152.4, 141.1, 127.5, and 98.5) were evident; thus compound 1 was bicyclic. The <sup>1</sup>H NMR spectrum showed proton resonances which could be assigned to an olefinic methine, an exomethylene group, three oxygen or nitrogen bearing methines, and an allylic methylene group. In addition, a methoxyl group and two exchangeable protons were evident. The deshielded nature of both the olefinic ( $\delta$  7.09) and exomethylene ( $\delta$  4.99 and 5.85) proton signals indicated that they each were  $\beta$  to carbonyl groups. It was possible to trace all of the proton-proton spin systems in 1 with data from a COSY-45 experiment. In the COSY spectrum, the oxymethine proton at  $\delta$  4.55 was coupled to the exchangeable OH proton at  $\delta$  7.67, to the olefinic proton at  $\delta$  7.09, and to a methine resonance at  $\delta$  3.83. The latter proton exhibited additional correlations with a methine at  $\delta$  4.11, which in turn was coupled to a pair of allylic methylene protons ( $\delta$  2.52 and 3.04). Both of these methylene protons also showed long range couplings with the olefinic proton. These data allowed assignment of partial structure 1a (Fig. 1). Further structural assignments were aided by heteronuclear correlation experiments. HSQC data indicated that the protons at  $\delta$  3.83 ( $\delta_c$  59.0) and 4.11 ( $\delta_c$ 72.6) were attached to carbons bearing nitrogen, and oxygen, respectively. In the HMBC spectrum, both of the methylene protons showed correlations to the olefin carbons at  $\delta$  127.5 and  $\delta$  141.1, while the olefinic proton showed a three bond correlation to the methylene carbon at  $\delta$ 30.2. This indicated that partial structure 1a was cyclized to provide substructure 1b. The methoxyl protons at  $\delta$  3.66 revealed a <sup>3</sup>J correlation with the carbonyl group at  $\delta$  166.3, which also exhibited HMBC correlations with the olefinic and methylene protons. These correlations defined the location of the methyl ester and helped establish the extended partial structure 1c.

<sup>&</sup>lt;sup>‡</sup> Compound 1: amorphous white solid;  $[\alpha]_D - 33.5$  (*c* 0.02, MeOH); UV (EtOH)  $\lambda_{max}$  (log  $\varepsilon$ ) 205 (4.08), 218 (4.14) nm; IR  $\nu_{max}$  (film) 3407, 1724, 1667, 1651, 1616, 1440, 1410, 1387, 1340, 1306, 1258, 1236, 1143, 1102, 1056, 1003, 967, 882, 785, 737 cm<sup>-1</sup>; <sup>1</sup>H NMR (500 MHz, C<sub>5</sub>D<sub>5</sub>N)  $\delta$  10.05 (1H, bs, NH), 7.67 (1H, d, J = 6.5 Hz, OH-6 $\alpha$ ), 7.09 (1H, bs, H-7), 5.85 (1H, bs, H-11a), 4.99 (1H, bs, H-11b), 4.55 (1H, m, H-6 $\beta$ ), 4.11 (1H, ddd, J = 16.5, 10.0, 6.0 Hz, H-10 $\beta$ ), 3.83 (1H, *t*, J = 10.0 Hz, H-5 $\alpha$ ), 3.66 (3H, s, OCH<sub>3</sub>), 3.04 (1H, dd, J = 17.5, 6.0 Hz, H-9 $\beta$ ), 2.52 (1H, m, H-9 $\alpha$ ); <sup>13</sup>C NMR (125 MHz, C<sub>5</sub>D<sub>5</sub>N)  $\delta$  166.3 (s, C-12), 160.4 (s, C-3), 152.4 (s, C-2), 141.1 (d, C-7), 127.5 (s, C-8), 98.5 (t, C-11), 72.6 (d, C-10), 70.8 (d, C-6), 59.0 (d, C-5), 51.9 (q, OCH<sub>3</sub>), 30.2 (t, C-9); CIMS *m*/*z* 240 [M+H]<sup>+</sup>, 257 [M+NH<sub>4</sub>]<sup>+</sup>; HR-FABMS: *m*/*z* 240.0781 [M+H]<sup>+</sup>; C<sub>11</sub>H<sub>14</sub>NO<sub>5</sub> requires: 240.0872.

<sup>&</sup>lt;sup>§</sup> Compound **2**: amorphous white solid;  $[\alpha]_D$  +53.0 (*c* 0.02, MeOH); UV (EtOH)  $\lambda_{max}$  (log ε) 207 (4.06), 217 (4.01) nm; IR  $\nu_{max}$  (film) 3515, 3216, 1718, 1679, 1630, 1432, 1374, 1300, 1256, 1238, 1110, 1062, 1049, 1009, 943, 899, 873, 785, 728 cm<sup>-1</sup>; <sup>1</sup>H NMR (500 MHz, C<sub>5</sub>D<sub>5</sub>N) δ 9.73 (1H, bs, NH), 7.50 (1H, bs, OH-10α), 6.85 (1H, bs, H-7), 5.82 (1H, d, J = 1.0 Hz, H-11a), 4.97 (1H, bs, H-11b), 4.59 (1H, m, H-6β), 4.02 (1H, m, H-10β), 3.71 (1H, *t*, J = 10.0 Hz, H-5α), 3.61 (3H, s, OCH<sub>3</sub>), 3.02 (1H, dd, J = 17.5, 5.0 Hz, H-9β), 2.53 (1H, m, H-9α); <sup>13</sup>C NMR (125 MHz, C<sub>5</sub>D<sub>5</sub>N) δ 166.2 (s, C-12), 161.0 (s, C-3), 153.1 (s, C-2), 134.4 (d, C-7), 130.9 (s, C-8), 98.7 (t, C-11), 75.7 (d, C-6), 67.0 (d, C-10), 58.1 (d, C-5), 52.0 (q, OCH<sub>3</sub>), 35.0 (t, C-9); HR-FABMS: *m/z* 240.0865 [M+H]<sup>+</sup>; C<sub>11</sub>H<sub>14</sub>NO<sub>5</sub> requires: 240.0872.



Figure 1. Partial structures of monocillinol A (1) and key HMBC correlations

The remaining structural elements of **1** consisted of an  $\alpha$ , $\beta$ -unsaturated amide carbonyl, an exomethylene group and one additional ring. The <sup>13</sup>C chemical shifts of the exomethylene group were highly polarized ( $\delta$  152.4 and 98.5), which indicated that it was bound to an oxygen. An HMBC correlation from H-10 to C-2 revealed that the C-10 oxygen substituent was linked to the exomethylene group. HMBC correlations from the amide NH to C-2, C-5, C-6, and C-10 (Fig. 1) then allowed assignment of the bicyclic structure shown for **1**. Comparison of the <sup>1</sup>H and <sup>13</sup>C NMR data reported for the substituted morpholine ring in compound **3**<sup>7</sup> with NMR data from the corresponding ring in monocillinol A (**1**) supported this structure. The relative stereochemistry in **1** was defined by 1,3-diaxial NOE interactions, which were observed in selective 1-D NOESY experiments, between H-5 and H-9 $\alpha$ , and between H-6 and the H-10 bridgehead proton (Fig. 2). While the structure of **1** is novel, it shares some features with the bicyclic ring system found in the chromophore of the proteinaceous antibiotic C-1027.<sup>8–10</sup>

The molecular formula of monocillinol B (2) was determined by HR-FABMS measurements to be isomeric with 1. The <sup>1</sup>H NMR spectrum of 2 was almost superimposable with that recorded for 1, and COSY data for 2 defined proton-proton spin systems which were similar to those



Figure 2. Key NOESY interactions observed in 1 and 2

observed in 1. The <sup>13</sup>C NMR spectral data of 1 and 2 corresponded closely; however, significant spectral differences were observed for the C-7 ( $\delta$  134.4) and C-8 ( $\delta$  130.9) olefin carbon resonances as well as the C-6 methine ( $\delta$  75.7) in compound 2.¶ In addition, the C-10 oxymethine carbon in 2 was also shifted upfield to  $\delta$  67.0. HMBC correlations established that the fused morpholine ring in monocillinol B (2) was cyclized via the oxygen substituent on C-6, as opposed to the C-10 oxygen in monocillinol A (1). Key data included a modest <sup>3</sup>J correlation between C-2 and H-6, and a <sup>4</sup>J correlation from one of the exomethylene protons ( $\delta$  5.82) to C-6. The amide NH ( $\delta$  9.73) also revealed correlations to C-2, C-5, C-6, and C-10. All of the other observed HMBC correlations were fully consistent with structure 2. The relative stereochemistry in 2 was assigned from NOE interactions (Fig. 2) in an analogous manner with 1.

Monocillinols A (1) and B (2) did not exhibit anti-HIV activity in an in vitro XTT-based assay<sup>11</sup> or cytotoxicity towards two human cell lines (LOX and OVCAR-3) at concentrations of 50  $\mu$ g/mL.

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 $<sup>\</sup>P$  To aid in spectral comparisons, the numbering for monocillinol B (2) is based on the same numbering scheme used for monocillinol A (1).